

## **Kimyoviy bog'lanish**

### **Test 1**

1. Bir xil metalmaslar orasida qanday kimyoviy bog'lanish hosil bo'ladi.  
1) ion; 2) vodorod; 3) qutbli kovalent; 4) qutbsiz kovalent

2. Quyidagi qaysi birikmada ion bog' mavjud.  
1)  $\text{H}_2\text{S0}_4$  2)  $\text{CuS0}_4$  3)  $\text{CCl}_4$  4)  $\text{HOH}$

3. Quyidagi qaysi moddada ion bog' mavjud emas  
1)  $\text{H}_2\text{O}_2$  2)  $\text{Na}_2\text{O}$  3)  $\text{NH}_4\text{Cl}$  4)  $\text{NaOH}$

4. CO molekulasi qanday bog'lardan tashkil topgan  
1)  $\delta$  – bog'lar 2)  $\pi$  – bog'lar  
3) bitta  $\delta$  – va bitta  $\pi$  – bog' 4) bitta  $\delta$  – va ikkita  $\pi$  – bog'

5. Qutbli kovalent bog'larga ega qutbsiz molekulani ko'rsating  
1) HCT 2)  $\text{BF}_3$  3)  $\text{Cl}_2$  4)  $\text{H}_2\text{O}_2$

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### **Test 2**

1. Turli xil metalmaslar orasida qanday bog' hosil bo'ladi  
1) qutbsiz kovalent; 2) ion; 3) vodorod; 4) qutbli kovalent

2. Quyidagi qaysi molekula qutbli kovalent bog'lanishga ega.  
1)  $\text{CO}_2$  2) KI 3)  $\text{F}_2$  4)  $\text{H}_2$

3. Quyidagi molekulalarning qaysi birida atomlar orasida vodorod bog'lanish hosil bo'ladi. 1)  $\text{HCHO}$  2)  $\text{C}_2\text{H}_5\text{OH}$  3)  $\text{CH}_3\text{OCH}_3$  4)  $\text{CH}_3\text{COCH}_3$

4.  $\text{SO}_2$  molekulasida S va O atomlari orasida qanday bog'lar mavjud  
1)  $\delta$  – bog'; 2)  $\pi$  – bog'; 3) bitta  $\delta$  – va bitta  $\pi$  – bog' 4) bitta  $\delta$  - va ikkita  $\pi$  – bog'

5. Qutbli kovalent bog'li, qutbsiz molekulani ko'rsating  
1)  $\text{BeF}_2$  2)  $\text{H}_2$  3) CO 4)  $\text{CH}_2\text{Cl}_2$

### **Test 3**

1. Faolmas metal va faolmas o'rtasida qanday bog' hosil bo'ladi.  
1) qutbli kovalent; 2) ion; 3) metal 4) qutbsiz kovalent
  
  2. Quyidagi qaysi birikma metal bog'lanishga ega  
1) FeO 2) FeC<sub>1</sub><sub>3</sub> 3) Fe 4) Fe(OH)<sub>3</sub>
  
  3. Qutbli kovalent bog'lanish yo'q birikmani aniqlang/  
1) H<sub>2</sub>S0<sub>4</sub> 2) Na<sub>2</sub>S0<sub>4</sub> 3) Na<sub>2</sub>S0<sub>3</sub> 4) Na<sub>2</sub>S
  
  4. В молекуле BF<sub>3</sub> molekulasida qanday bog'lar bor.  
1)  $\pi$  – bog' 2)  $\delta$  – bog'  
3) bittab  $\delta$  – va bitta  $\pi$  – bog' 4) bitta  $\delta$  – va ikkita  $\pi$  – bog'
  
  5. Qutbli kovalent bog'li qutbsiz molekulani aniqlang.  
1) CO 2) F<sub>2</sub> 3) CH<sub>4</sub> 4) H<sub>2</sub>S
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### **Test 4**

1. Faol metallar metallar va faol metalmaslar orasida qanday bog'lanish hosil bo'ladi  
1) qutbli kovalent; 2) ion; 3) metal; 4) qutbsiz kovalent bog'lanish
  
2. Quyidagi qasi molekula qutbsiz kovalent bog'lanishga ega,  
1) HC1 2) NaC1 3) SiF<sub>4</sub> 4) O<sub>2</sub>
  
3. Quyidagi qaysi molekulada vodorod bog'lanish hosil bo'lmaydi.  
1) H<sub>2</sub>O 2) HF 3) H<sub>2</sub>S 4) HCOOH
  
4. H<sub>2</sub>O<sub>2</sub> molekulasida faqat ..... bog'lar bor.  
1)  $\delta$  – bog'; 2)  $\pi$  – bog'; 3) bitta  $\delta$  – va bitta  $\pi$  – bog'; 4) bitta  $\delta$  - va ikkita  $\pi$  bog'
  
5. Qutbli kovalent bog'lanishli qutbsiz molekulani ko'sating  
1) O<sub>2</sub> 2) HC1 3) S0<sub>2</sub> 4) C0<sub>2</sub>

## **Tes<sup>t</sup> 5**

1. Quyida keltirilgan qaysi element atomi o'zidan electron berib musbat zaryadlangan ionga aylanadi:

- a) C      b) Li      s) Ba      d) P

2. ion bog'lanishli modda:

- a) SO<sub>2</sub>      b) KC1      s) BaC1<sub>2</sub>      d) NH<sub>3</sub>.

3. Molekulyar Kristal panjaraga ega modda:

- a)HF      b)H<sub>2</sub>      s)LiF      d) BaC1<sub>2</sub>.

4. Qaysi moddaning suyuqlanish harorati yuqori:

- a) LiF      b) HI      s) I<sub>2</sub>      d) HCl .

5. Quyidagi qaysi birikmada uglerodning oksidlanish darajasi +4 ga teng:

- a) Na<sub>2</sub>CO<sub>3</sub>      b) CH<sub>4</sub>      s) C0<sub>2</sub>      d) SiC.
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## **Tes<sup>t</sup> 6**

1. Quyida keltirilgan qaysi element atomi o'zidan electron qabul qilib manfiy zaryadlangan ionga aylanadi:

- a)Ca      b)Na      s)Br      d)F

2. Qutbli kovalent bog'lanishga ega modda:

- a) P<sub>2</sub>O<sub>5</sub>      b) KF      s) BaF<sub>2</sub>      d)H<sub>2</sub>S.

3. Ion Kristal panjaraga ega modda:

- a)KF      b)CsCl      s) HI      d)C1<sub>2</sub>.

4. Quyidagi vqaysi moddaning suyuqlanish harorati yuqori:

- a)KF      b)HBr      s)BR<sub>2</sub>      d)H<sub>2</sub>O.

5. Quyidagi qaysi moddada uglerodning oksidlanish darajasi - 4 ga teng:

- a) CO<sub>2</sub>      b) CC1<sub>4</sub>      s) CH<sub>4</sub>      d) H<sub>2</sub>CO<sub>3</sub>.